Dhaka Board 2017

Chemistry

Subject Code | 1 | 3 | 7

Time — 2 hours 35 minutes

Creative Essay Type Questions

Full marks - 50

[N.B.-The figures in the right margin indicate full marks. Read the stems attentively and answer any five of the following questions.]

1.

Period	Gr I	2	-	14	17
2	Li	_	-	С	_
3	Х	Y	-	-	Cl
4	Z	_	-	N .=	_

[Here x, y, z are used as symbol, they are not the symbol of any regular elements.]

What is "Stoichiometry"? a.

Explain NH₃ is basic in nature. b.

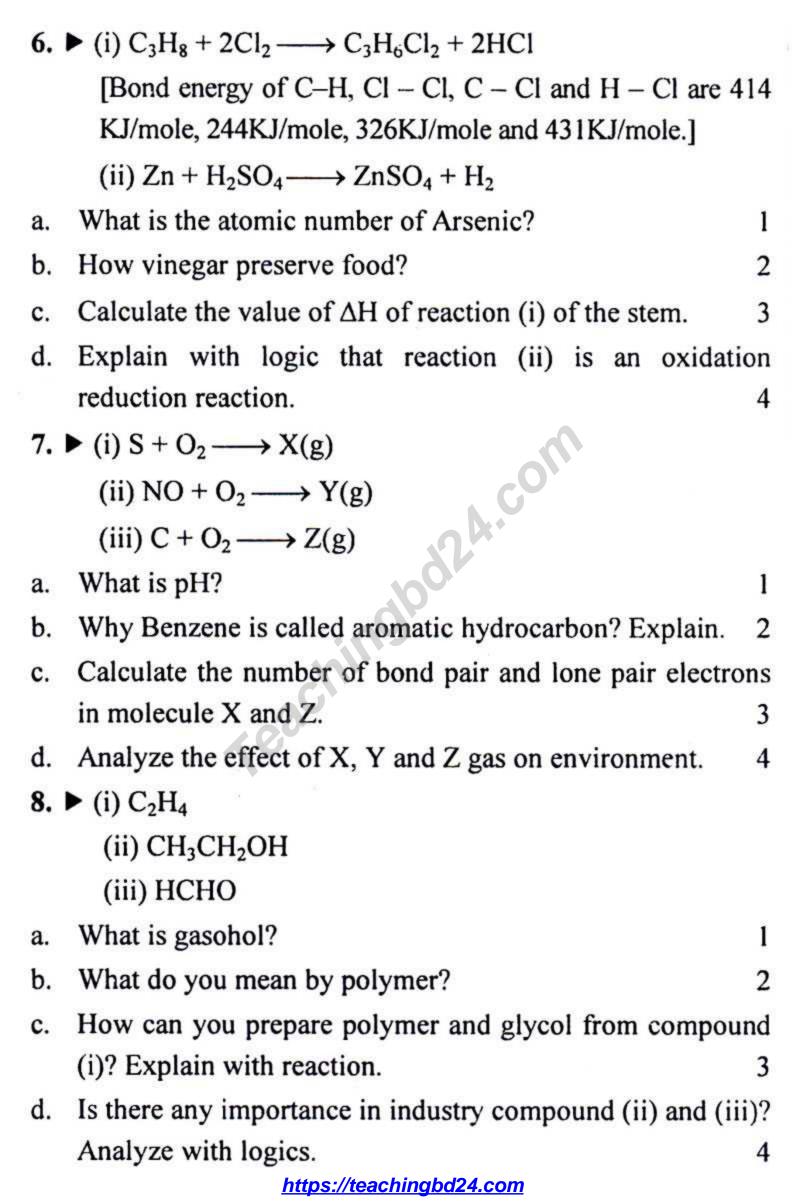
Explain the order of atomic size of Gr-I and period-3 C. elements of the stem.

- Compound of Y and Cl and compound of C and Cl, which d. one conduct electricity and why explain?
- 2. ▶ Being separated CaCO₃ by heat "B" and "C" gas (mass-44) are produced. Compound "D" is got by adding water with "B" and "E" a compound can be found by throwing Cl2 gas within "D".
- What is saponification? a.

Define the role of baking powder to indigation. b.

"C" is an acidic gas—Explain. C.

d.	Compound "E" is an effective germ killer — Explain it.	4
3.	▶ "X" and "Y" two atoms whose atomic number are 2	20
and	117:-	
a.	What is radical?	1
b.	Explain valency and valence electron.	2
c.	Explain with diagram formation of "Y2" molecule.	3
d.	Is the compound formed by "X" and "Y" is soluble	in
	water?— Explain with logic.	4
4.	► To prepare 40.5g Ca (HCO ₃) ₂ mixed 25g CaCO ₃ , 4.5	ig
H ₂ (O and 8g CO ₂ . But does not get expected product.	
a.	What is the value of bond energy of O-H?	1
b.	What is meant by galvanizing?	2
c.	How much 'mole' CO ₂ used in the reaction?	3
d.	Explain the cause of getting less product in the reaction.	4
5.	(i) Mg (OH) ₂ + 2HCl \longrightarrow MgCl ₂ + 2H ₂ O	
	(ii) $CaO + CO_2 \longrightarrow CaCO_3$	
	(iii) $Ca + F_2 \longrightarrow CaF_2$	
a.	What is oxidation number?	1
b.	Why concentrated HNO ₃ kept in brown colored bottle	e?
	Explain.	2
c.	Explain the role of like (i) reaction in our daily life.	3
d.	All synthesis reactions are addition but all addition reaction	ns
	are not synthesis — Analyze depend on (ii) and (iii) no. https://teachingbd24.com	4



Creative Multiple Choice Questions

Subject Code 1 3 7
Full marks — 25

[NB. Answer all the questions. Each question carries one mark. Block fully, with a ball point pen, the circle of the letter that stands for the correct/best answer in the "Answer Sheet" for the Multiple Choice Questions Examination.]

1.	What	is	the	standard	pH	value	for
	skin o	f h	uma	n body?			

- **a** 5.0
- **ⓑ** 5.5
- © 7.0
- d 7.8

2. Which one diffuse more?

- @ He
- **ⓑ** H₂
- © CO₂
- @ NH₃

3. Which one is the last step of investigation and research method?

- (a) Taking advance idea about result
- (b) Effect on science and welfare of human being
- © Gaining knowledge about the subject
- (d) Consisting and analyze about the information

Follow the stem and answer questions no. 4 and 5:-

 $CH_3 - CH = CH_2 + R(s) + H_2O$

4. What is R?

- (a) Alcohol
- (b) Acid
- © Base
- @ Salt

5. Alkene of the stem-

- i. catalytic hydrozination is possible
- Potassium permanganet is obtain by oxidation
- iii. polymerisation is possible

Which one is correct?

@ ii

- · (b) i and iii
- © ii and iii
- (d) i, ii and iii

6. With how many elements Lavosieya

made periodic table?

- (a) 33
- **ⓑ** 63
- © .67

@ 92

7. Which gas is volatile at solid state?

- @ CO
- **ⓑ** CO₂
- © NO₂
- @ NH₃

8. Which one is the ore of lead?

- (a) Limonite
- (b) Hematite
- © Megnatite
- d Galena

9. How many molecules of water are there in blue vitriol?

@ 1

ⓑ 3

© 5

@ 7

10. Which one is known as bio fuel?

- @ CH4
- (b) CH₃OH
- © CH₃CH₂OH
- ⊕ CH₃COOH

11. What is the symbol of potassium?

(a) P

(b) Po

© K

Ka

12. How many amino acid are present in insuline polymer?

- @ 20
- **ⓑ** 21
- © 22

@ 23

13. In which oxidation is possible?

- (a) Fe2+
- ⓑ Cu²⁺
- © Fe3+
- @ Na⁺

- @ 0.10M
- **ⓑ** 0.37M
- © 1.00M
- @ 2.65M

15. N	Monomer of P	VC is—
i.	polypropane	
ii	i. vinyl chloric	ie
ii	ii. monochloro	ethene
V	Which one is co	orrect?
(8	i and ii	(b) i and iii

16. Which one is used in treatment of blood Leukemia?

@ ²³⁸Pu

© ii and iii

(b) 131 I

d i, ii and iii

© 106Ru

(d) ^{32}P

Follow the stem and answer questions no., 17 and 18:-

Element	Period	Group			
P	2	2			
Q	3	17			
R	4	2			

[Here P, Q, R are the symbolic terms]

17. In which solvent dissolve the compound formed by R and Q?

- Alcohol
- Water
- © Kerosene
- d Carbon tetrachloride

18. ☐ In case of the order of P, Q and R element—

i. Electro negativity: Q > R

ii. Atomic size: R > Q > P

iii. Ionization potential: R > P

Which one is correct?

(a) i and ii

(b) i and iii

© ii and iii

(d) i, ii and iii

19. In which Copper slag is dissolved?

(a) In water

⑤ In cytric acid

© In alcohol

d In carbon tetrachloride

20. Which elements are there in industrial grade CaC₂?

@ As, N

© As, S

d As, P

21. How many lone pair of electrons are there in PCl₃?

@ 0

6 1

© 3

@ 10

22. ★ 14R is an element—

- oxide of the above element has high boiling and melting point
- ii. group number is four in the periodic table

iii. it form ionic bond with bromine Which one is correct?

(a) i

Ъ ii

© i and iii

d i, ii and iii

23. What will be the colour if wet blue litmus paper pass through SO₃ gas?

(a) Green

(b) Red

© Colourless

d Orange

24. Which one is an electronic conductor?

(b) Alkali solution

© Acid solution

d Salt solution

25. What is the oxidation number of Sulphur in potassium sulphide?

(a) - 2

(b) 0

©+4

(d) + 6

ž.	1	(b)	2	(6)	3	(b)	4	(1)	5	®	6	3	7	6	8	@	9	©	10	©	11	©	12 25	©	13	(8)
Ar	14	©	15	©	16	@	17	®	18	3	19	6	20	@	21	6	22	3	23	(b)	24	3	25	3	191	